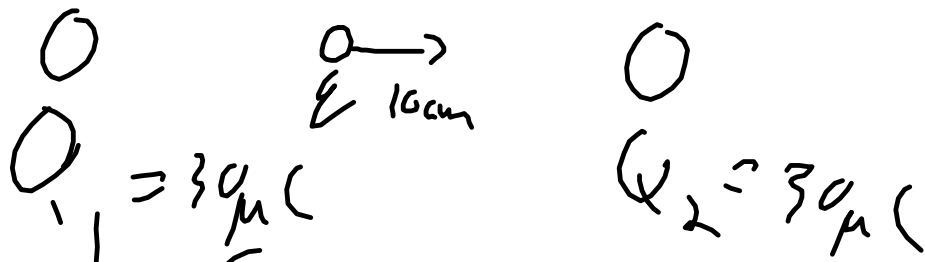


$$\% \text{ error} = \frac{|K_1 - K_2|}{\frac{K_1 + K_2}{2}} \times 100\%$$

$$\frac{K_1}{V_a} = \text{Slope Part 1} \quad K_2 V_a = \text{Slope Part 2}$$

P456 Q13 $\xleftrightarrow[0.20 \mu\text{C}]{6.0 \text{ m}}$



~~$W = Fd$~~
Not constant

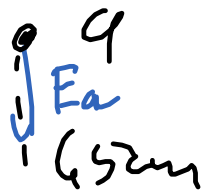
$$W = \Delta E_e = \left(\frac{K Q_1 q}{r_1} + \frac{K Q_2 q}{r_2} \right)$$

$$- \left(\frac{KQ_1q}{r_{i1}} + \frac{KQ_2q}{r_{i2}} \right)$$

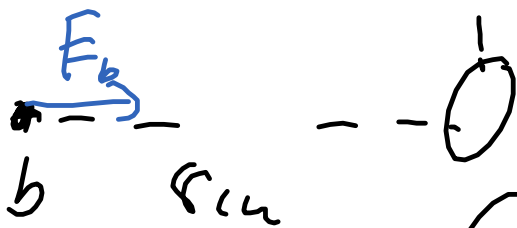
$$KQq \left[\left(\frac{1}{r_{f1}} + \frac{1}{r_{f2}} \right) - \left(\frac{1}{r_{i1}} + \frac{1}{r_{i2}} \right) \right]$$

$$9 \times 10^9 (30 \times 10^{-6}) (0.2 \times 10^{-6}) \left[\left(\frac{1}{0.2\text{m}} + \frac{1}{0.4\text{m}} \right) - \left(\frac{1}{0.3\text{m}} + \frac{1}{0.3\text{m}} \right) \right]$$

Q15

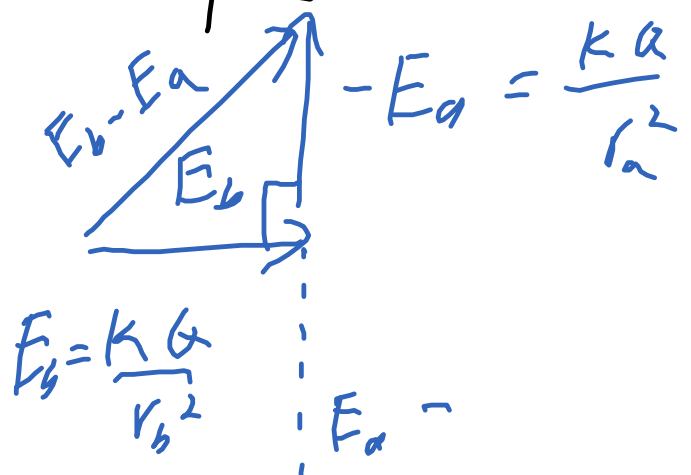


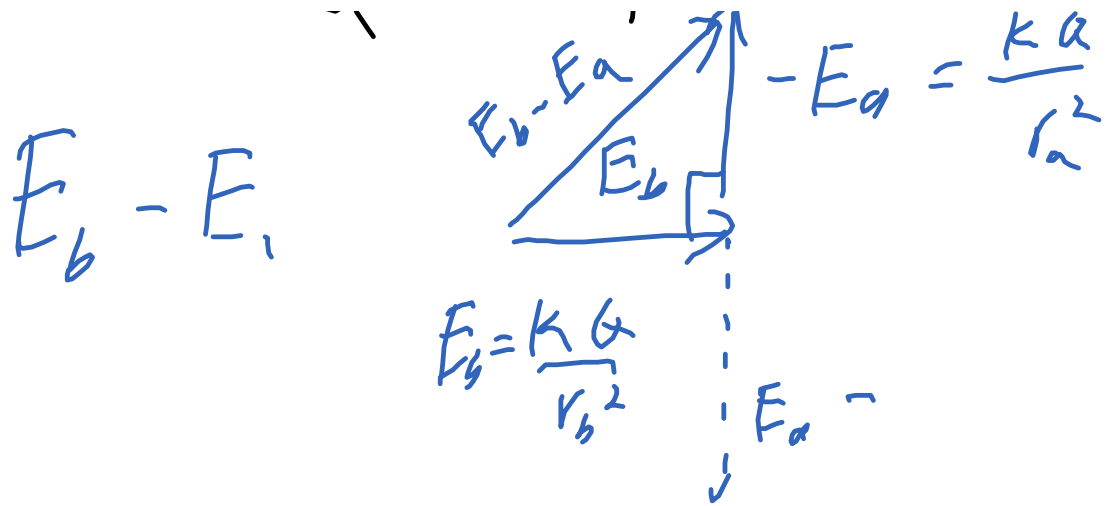
$$a) KQ \left(\frac{1}{r_+} - \frac{1}{r_-} \right)$$



$$Q = -2.8 \mu\text{C}$$

$$E_b - E_a$$





Electric Current, I

Galvani - stuck metals into dead frogs, the legs twitch. Thought he had "life force".

Volta - 2 different metals in a electrolyte solution (eg. HCl solution) the metals have a potential difference - continues even as you discharge the terminals.

You can get a constant flow of charge.

Rate of flow of charge is called current, I.

$$I = \Delta q/t$$

Units: ampère, $A = C/s$

Homework:

P476 Q1-3

P477 Problems 1-9 odds

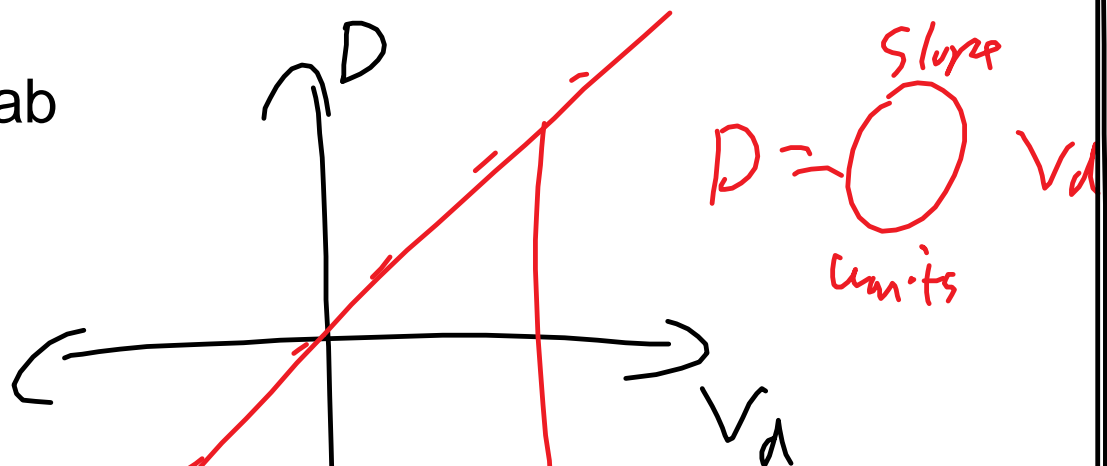
Lab due next class:

Purpose, hypothesis (include diagram and notes from class), procedure - say "refer to lab manual p75", data tables with units, graphs with equations and units and variables, % error calculation, conclusion, sources of uncertainty

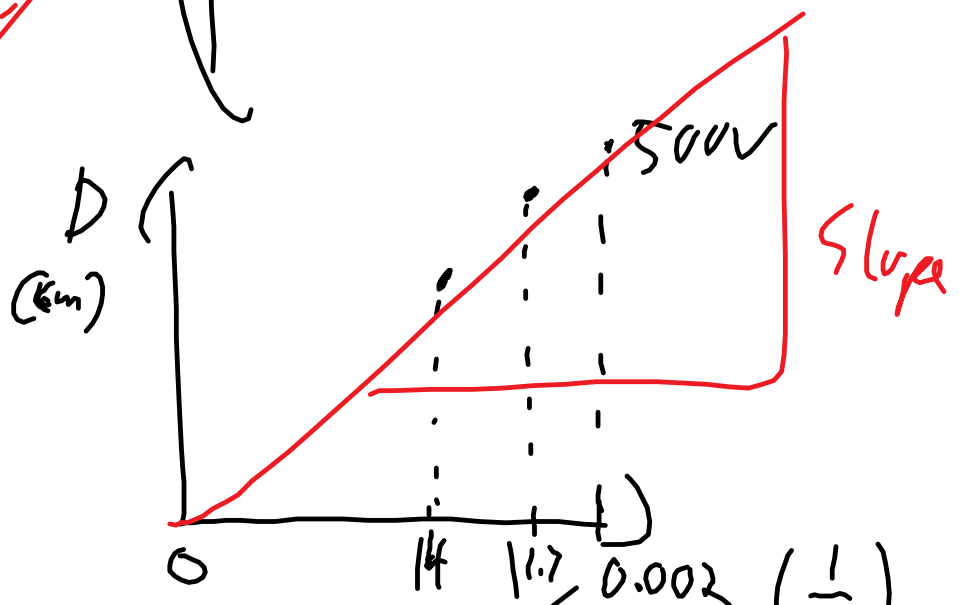
Field trip form.

Block 1-1 Lab

Part 1



Part 2



$$D = \frac{51.4}{V_a} \quad \frac{1}{V_a}$$

0 14 11.7 0.002 ($\frac{1}{V}$)

$$d/y =$$

P456

$$V = ? \quad E_k = 38 \text{ keV}$$

$$Q = 3.2 \times 10^{-19} \text{ C} = 2e$$

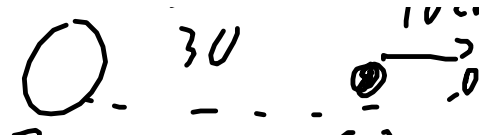
$$V = \frac{E_e}{q} \quad \Delta E_k = \Delta E_e$$

$$E_k = Vq$$

$$V = \frac{38000 \text{ eV}}{2e} = 19 \text{ kV}$$

Q 15

$$\rightarrow \frac{60 \text{ cm}}{10 \text{ cm}}$$


 $Q_1 = 30 \mu C$ $q = 0.2 \mu C$ $30 \mu C = Q_2$
 $W = ?$

$W = \cancel{F d} = q \cancel{E d}$
 not constant

$$W = \Delta E_e = E_{ef} - E_{ei}$$

$$E_{ef} = \left(\frac{K Q_1 q}{r_{1f}} + \frac{K Q_2 q}{r_{2f}} \right)$$

$$- \left(\frac{K Q_1 q}{r_{1i}} + \frac{K Q_2 q}{r_{2i}} \right)$$

$$K Q q \left[\left(\frac{1}{r_{1f}} + \frac{1}{r_{2f}} \right) - \left(\frac{1}{r_{1i}} + \frac{1}{r_{2i}} \right) \right]$$

$$9 \times 10^9 (30 \times 10^{-6}) (0.2 \times 10^{-6}) \left[\left(\frac{1}{0.4} + \frac{1}{0.2} \right) - \left(\frac{1}{0.3} + \frac{1}{0.3} \right) \right]$$

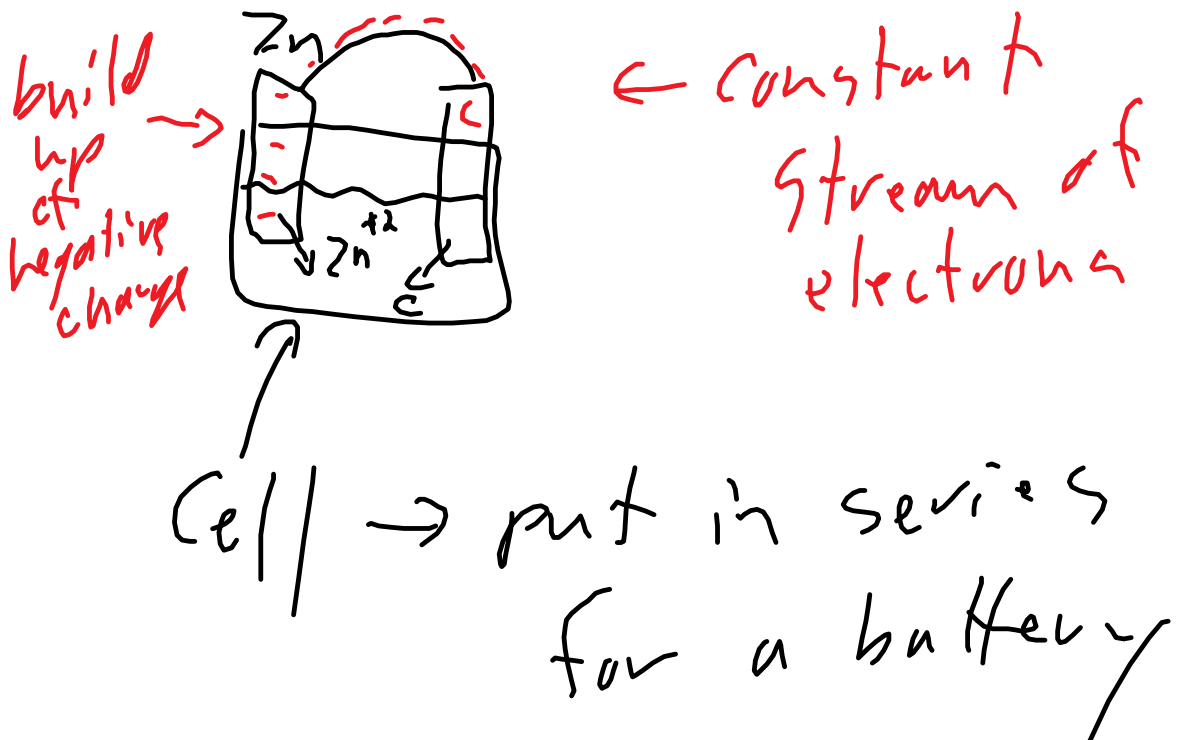
$$9 \times 10^9 \left(30 \times 10^{-6} \right) \left(0.2 \times 10^{-6} \right) \left(0.4 \quad 0.2 \quad 0.2 \quad 0.2 \right)$$

Electric Currents, I

Galvani - iron spike in the spinal cord of a frog on a copper plate, the legs would twitch and jump even without a brain.

Galvani thought he had discovered the "life force". It didn't work if you used the same metal.

Volta - If you put different metals into a electrolyte solution (eg. HCl) there is a potential difference between the metals.



The rate of flow of charge is called the current, I.

$$I = \Delta q / \Delta t$$

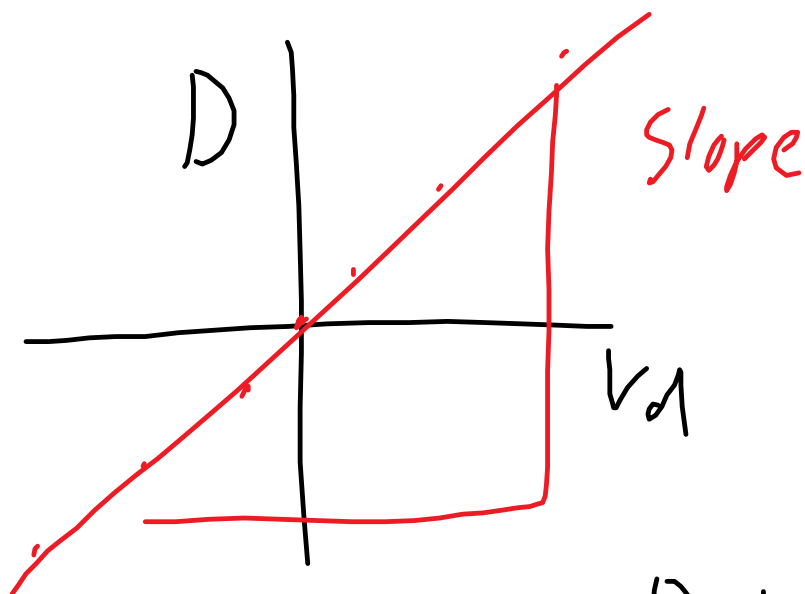
Units: ampère, 1 A = 1 C/s

Lab due Friday

Questions HW

Quiz Tuesday: Voltage and energy

Electrical Current



$$D \propto V_d$$

$$D \propto \frac{1}{V_a}$$

$$D = K \frac{V_d}{V_a}$$

next 1

Part 1 $D = \left(\frac{K_1}{V_d} \right)$ \swarrow slope \nearrow Part 1

Part 2 $D = \left(K_2 V_d \right) \frac{1}{V_d}$ \swarrow slope \nearrow Part 2 $V_d = 5$ materials

$$\% \text{ error} = \frac{|K_1 - K_2|}{\frac{K_1 + K_2}{2}} \times 100\%$$

Current Electricity

Story:

Galvani- Stuck dead frogs with iron and copper and they jumped. Thought he had discovered the life-force.

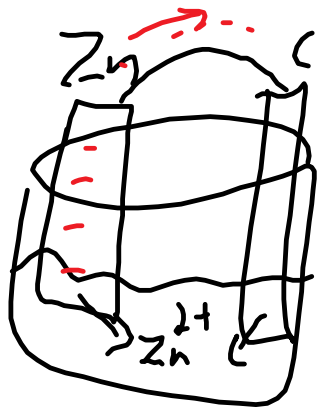
Volta - It isn't the life-force, it is a property of the metal and the solution (frog juices).

Two metals into an electrolyte solution (eg. HCl) they can have a potential difference.

Copper and aluminum $V = 0.4V$

Zinc and Carbon $V = 1.4 \text{ V}$

What's the deal?



the Zinc goes
into solution
leaving a charge
on the terminal.

electrochemical cell

- the current continues
even as the terminal
is discharged,

Current, I is the rate of transfer of charge, q .

$$I = \Delta q / \Delta t$$

Units: ampère, A $1A = 1C/s$